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# Effects of Contaminated St. Lucie River Saltwater Sediments on an Amphipod (Ampelisca abdita) and a Hard-Shell Clam (Mercenaria mercenaria)

Tham C. Hoang Loyola University Chicago, thoang@luc.edu

Gary M. Rand Florida International University

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1	Effects of St. Lucie River (FL) saltwater sediments on the amphipod (Ampelisca abdita) and the hard
2	shell clam (Mercenaria mercenaria)
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4	Tham C. Hoang <sup>1,2</sup> , Gary M. Rand <sup>1*</sup>
5	<sup>1</sup> Florida International University, Southeast Environmental Research Center, Earth & Environment
6	Department, North Miami, Florida, USA
7	<sup>2</sup> Present address: Institute of Environmental Sustainability,
8	Loyola University Chicago, Chicago, Illinois, USA
9	
10	*Corresponding author
11	Dr. Gary M. Rand
12	Ecotoxicology and Risk Assessment Laboratory
13	Southeast Environmental Research Center
14	Florida International University
15	North Miami Beach, FL
16	Email: <u>randg@fiu.edu</u>
17	

## 18 Abstract

The St. Lucie estuary ecosystem in South Florida has been noted to be contaminated with 19 metals and pesticides. Our earlier studies showed that aquatic organisms, especially benthic species in 20 21 the St. Lucie estuarine ecosystem are at high risk of copper (Cu) exposures. The objectives of this study are to conduct tests with separate groups of organisms exposed to 7 field-collected sediment 22 23 samples from the St. Lucie River according to standard procedures to evaluate toxicity and tissue concentrations of Cu and zinc (Zn). Short term and long term whole sediment acute toxicity tests were 24 conducted with Ampelisca abdita and Mercenaria mercenaria. Analysis of sediment chemical 25 26 characteristics showed that Cu and Zn are most concern because their concentrations in 86% of the sediments were higher than the threshold effect concentrations for Florida sediment quality assessment 27 and the NOAA SQuiRTsnational Cu sediment quality guidelines. There was no significant effect on 28 survival of the tested organisms. Elevated Cu and Zn concentrations in the test organisms were found. 29 Dry weight of the tested organisms was inversely related to Cu and Zn concentrations in sediments and 30 organisms. The effects on organism weight and Cu and Zn uptake raise a concern about the organism 31 population dynamics of the ecosystem because benthic organisms are primary food sources in the St. 32 Lucie system and are continuously exposed to the Cu and Zn contaminated sediments for their life 33 34 cycle. The present study also indicates that Cu and Zn exposures via sediment ingestion were more 35 important than pore water exposure.

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Key words: Cu uptake, Cu-contaminated sediment, St. Lucie River sediments, *Ampelisca abdita*, *Mercenaria mercenaria*

## 39 Introduction

The St. Lucie Estuary (SLE) watershed is composed of five major drainage basins and several 40 smaller basins in the northern portion of St. Lucie County, Florida. It contains the most concentrated 41 42 citrus agriculture acreage in South Florida. The SLE is located on the Martin/St. Lucie County line. The inner SLE is comprised of the North Fork and the South Fork of the St. Lucie River (SLR) and has a 43 total surface area of 6.4 square miles. The two forks converge to form a single middle estuary with a 44 surface area of 4.7 square miles. The middle estuary extends east from approximately 5 miles until it 45 meets the Indian River Lagoon, which opens to the Atlantic Ocean at the St. Lucie Inlet. A heavier 46 47 concentration of citrus agriculture ( $\sim 60\%$ ) land use potentially affects the drainage basins into the North Fork compared to the South Fork (~45%) of the SLR. In 1972, the Florida Trustees recognized the 48 ecological importance of the North Fork of the St. Lucie River by designating it an Aquatic 49 Preserve/Outstanding Florida Water. 50

Copper has a long history of use in agriculture (e.g., citrus groves) as a fungicide and fertilizer in 51 south Florida (Alva et al. 1995). In the early 1900's copper containing fertilizers for citrus groves 52 accounted for as much as 34 kg Cu/ha annually and fungicidal sprays contributed an additional 10kg 53 Cu/ha annually. Surface soils (0-15 cm) for mature citrus groves contained as much as 540kg Cu/ha. 54 55 Increased levels of copper in South Florida soils have been a result of repeated applications of copper over several decades to agricultural areas and soil copper concentrations increase proportional age of 56 citrus production (Reuther and Smith, 1952). According to the U.S. Department of Agriculture in 2005, 57 over 500,000 kg of copper (as copper hydroxide, copper sulfate or basic copper) were applied to 58 grapefruit, orange, tangelo, tangerine and temple crops on 259,563 ha in Florida (USDA 2006). These 59 quantities do not reflect use of Cu on other citrus crops or the use of copper sulfate and chelated Cu 60 61 formulations (e.g., Cutrine-Plus, Komeen, etc.) as algaecides-herbicides, which are permitted by the

Florida Department of Environmental Protection (FDEP) for control of nuisance planktonic and 62 filamentous algal and vascular plants (Leslie, 1990). Note that Cu also does leach from boats into both 63 fresh- and salt-water (e.g., harbors and marinas) because it is a component of antifouling pants. More 64 recently, statewide pesticide usage data (based on total lbs a.i. applied) compiled by the Florida 65 Department of Agriculture and Consumer Services (FDACs 2010) from 2007-2009 for 14 crops and 169 66 67 active ingredients (a.i.) ranked copper hydroxide (1,176,500 lbs. a.i. applied) number 5 out of 10 pesticides and the most applied fungicide (number 1 out of 10 fungicides). 68 A comparison of aqueous Cu concentrations in agriculture and non-agriculture watersheds shows 69 70 higher concentrations in runoff where agriculture was practiced compared to runoff near non-agriculture land (Dietrich et al., 2001). Copper loads in surface runoff are related to total Cu in soils, soil 71 properties, metal characteristics and environmental factors, especially in sandy soils (He et al., 2006). 72 Enrichment of Cu in runoff will adversely affect receiving surface water quality (Moore et al., 1998). 73 As a result of the use of copper in agriculture in St. Lucie County, it appears that concentrations of 74 copper (and other contaminants, such as Zn) from drift and/or from surface runoff of contaminated soils 75 (or soil erosion) may also produce exposures that adversely affect saltwater benthic communities, when 76 the Cu-contaminated soils reach and become incorporated as part of the sediments of the St. Lucie River 77 78 system. Sediment chemistry data indicate that Florida coastal sediments in several areas are contaminated with metals (Long and Morgan 1990, Delfino et al. 1991, FDEP 1994), especially Cu 79 (Haunert 1988, Trefry and Trocine 2011, Trocine and Trefry 1993, 1996). 80 81 Our early laboratory results indicate that copper-contaminated Florida agricultural soils that are flooded likely promote the release of Cu from soils producing adverse effects on freshwater organisms 82 83 (Hoang et al. 2008a, 2009a, b). In addition, we showed high potential ecological risks to aquatic species 84 as a result of Cu exposures in sediment and water and high probability of exceedences of the Florida

85 Department of Environmental Protection Sediment Quality Assessment Guideline values for the

86 Threshold Effect Concentration and the Probable Effect Concentration (FDEP SQAGs TECs and PECs)

for Cu (FDEP 2003) in the St. Lucie River (Schuler et al. 2008). More recently, Carriger and Rand 2013

88 (in press) also showed high ecological risks of Cu in this system to aquatic organisms. The objectives of

89 the present study are to conduct whole sediment toxicity studies with the clam (Mercenaria mercenaria)

90 and the benthic amphipod (*Ampelisca abdita*) exposed to field-collected sediment samples from the St.

91 Lucie River to evaluate uptake (bioconcentration) and toxicity of Cu and Zn.

92

#### 93 Materials and Methods

94 Sediments used in the present study (n=7; 6 test and 1 reference site) were collected by the

95 National Oceanic and Atmospheric Administration (NOAA) from the St. Lucie River, south Florida,

96 USA and transferred to the Ecotoxicology and Risk Assessment Laboratory (ERAL) of Florida

97 International University on April 5-7, 2011 for toxicity testing (Figure 1). ERAL is a NELAC-

98 accredited laboratory facility for fresh- and salt-water toxicity testing. Sediment samples were labeled

99 NOA2581 (reference site in the South Fork of the SLE) along with six test sites as NOA2569,

100 NOA2334, NOA2640, NOA2639, NOA 2333, and NOA2570 (sites in the North Fork of the SLE). Prior

101 to aquatic testing, sediments were physically characterized and background concentrations of metals and

102 organic pollutants were analyzed. The sediments were also analyzed for acid volatile sulfide (AVS) and

simultaneously extracted metals (SEM). Using the AVS/SEM ratio, we can predict the bioavailability of

104 metals (Berry et al. 1996).

105 Two separate types of studies were conducted with the 7 field-collected sediment samples to 106 evaluate mortality, growth and accumulation; one study with the tube-dwelling amphipod (*A. abdita*), 107 which is a common standard saltwater benthic test species used for whole-sediment toxicity and bioaccumulation tests and one study with the hard shell clam (*M. mercenaria*) which is a native species
in the St. Lucie system. Exposures to the field-collected sediments in both studies were in a flowthrough water system to ensure consistent water quality conditions (e.g., low ammonia concentrations).
Flow-through systems were calibrated prior to testing to ensure correct water placements in test
chambers over each 24-hour time period. Saltwater for the flow-through system was obtained from a
saltwater well (with Biscayne Bay water) which was air-sparged, carbon-filtered and UV-sterilized with
a salinity of 31ppt and a pH of 8.0-8.5.

Toxicity tests with A. abdita were 10 days in duration and were conducted according to the 115 116 methods for measuring the toxicity and bioaccumulation of sediment-associated contaminants with 117 marine invertebrates (U.S. EPA. 1994). A. abdita were obtained from a commercial supplier. An initial subsample of the A. *abdita* population was used for length and weight measurements and tissue analyses 118 119 (Cu, Zn) for background data. There were 10 organisms per replicate with 8 replicates per sediment sample site. Organisms (10) were randomly distributed in each test chamber with 350ml of water and 120 150 ml of sediment (8 test chambers /site; 80 organisms exposed/site) in the water bath of the flow-121 122 through water system (2 test chamber water volume turnovers/24h). Water quality monitoring for the tests included salinity, ammonia and pH measurements at test initiation and at test termination. 123 124 Temperature and dissolved oxygen were measured daily. Salinity, ammonia, and pH were measured using a YSI conductivity/salinity meter (YS Inc, Yellow Springs, Ohio, USA), an Accumet® Ammonia 125 Electrode (Fisher Scientific, Pittsburgh, PA, USA), and an Accumet pH meter (Vernon Hills, Illinois, 126 127 USA), respectively. Water temperature and dissolved oxygen concentrations were measured using a YSI dissolved oxygen meter (YS Inc., Yellow Springs, Ohio, USA). Mortality of A. abdita was measured at 128 test termination along with growth (dry weight and length) of surviving organisms. The quality criterion 129 130 for control survival was 80%. Tissue concentrations (whole body) of Cu and Zn were measured at test

initiation and again at test termination. To increase the detection limits for Cu and Zn, surviving

132 organisms from all replicates were combined for each treatment and digested with HNO<sub>3</sub> acid using a

Hotblock and based on the U.S. EPA Method 3050B (U.S. EPA 1996a) for tissue Cu and Zn analyses.

134 Analysis of Cu, Zn, and other minerals were conducted with an inductively coupled plasma

135 spectrometer (Thermo Scientific Inc. 5225 Verona Road, Madison, WI 53711).

136 In addition, a 28-day bioconcentration study was conducted similarly to U.S. EPA Ecological Effects Test Guidelines (U.S. EPA, 1996b) except that sediment was the source of the contaminants (not 137 water exposure). The hard shell clam (M. mercenaria), an economically important native species in the 138 139 SLR system was the test species and only a 28-day uptake phase (without a depuration phase) was used. *M. mercenaria* juveniles were obtained from a commercial supplier. An initial subsample of the *M*. 140 mercenaria juvenile population was used for weight measurements and tissue analyses of Cu and zinc. 141 There were 50 organisms per replicate with 2 replicates per sediment sample site used. Organisms (50) 142 were randomly distributed to an 18L test chamber with 12L of water and 3-4 cm of sediment (2 test 143 chambers /site; 100 organisms exposed/site) in a flow-through water system (4 tank water volume 144 turnovers/24h). Water quality monitoring for the test included salinity, conductivity, and pH 145 measurements for the first 3 days, daily and again at the end of the test. Temperature and dissolved 146 147 oxygen were measured daily. Mortality was measured on days 3, 7, 14, 21 and 28. Tissue samples (n=4) were also collected and measured on days 3, 7, 14, 21 and 28 of surviving organisms for Cu and Zn. 148 The quality criterion for control survival was 80%. Overlying and porewater samples were also 149 150 collected when the tissue samples were collected for analyses of Cu, Zn and dissolved organic carbon (DOC). Porewater was collected by centrifuging the sediments at 2500g and 4° for 30 minutes (Ankley 151 152 et al. 1991). The samples were filtered with 0.45 µm filters prior to analysis. DOC was analyzed with a 153 Shimadzu TOC-5000 (Shimadzu Scientific Instruments, Columbia, MD, USA). Measurement of water

quality and tissue Cu and Zn concentrations were conducted as described for the *A. abdita* test above.
Digestion of sediment, clam, and amphipod samples for analysis of Cu and Zn were conducted at
Loyola University Chicago.

At the end of the study, survival, dry weights, Cu and Zn tissue concentrations were analyzed to 157 determine whether there were statistically significant (p < 0.05) treatment-related effects (responses) of 158 the test substance. The ANOVA F-test was used to test the null hypothesis; that the effects of all 159 sediments including the reference sediment are the same. Tissue Cu and Zn concentrations at the end of 160 the tests or at time intervals during the test (*M. mercenaria* test only) were compared with initial Cu and 161 Zn concentrations (background) using Dunnett's procedure. Multiple correlations between dry weight, 162 tissue Cu and Zn concentrations, sediment Cu and Zn concentrations were conducted to determine 163 cause-effects (response) relationship. All statistical analysis was conducted using SAS (version 9.2). 164

165

#### 166 **Results and discussion**

#### 167 Sediment characteristics and chemistry

Characteristics of the sediments are shown in Table 1. In general, the cation exchange capacity was high which suggests that the sediments have high potential to retain metals. Results of AVS and total SEM are shown in Table 1. AVS for sediments NOA2333, NOA2334, NOA2569 and NOA2570 were below the detection limits. The total SEM/AVS ratios for these sediments were estimated based on the detection limits of AVS for those samples. The ratio of total SEM/AVS for all sediments was greater than 1 which suggests that metals in the sediments are more bioavailable to benthic organisms (Ankley et al.1996, McGrath et al. 2002).

Concentrations of metals and minerals in the sediments from the sites are shown in Table 2a.
Concentrations of the metals and minerals varied from site to site. Among the toxic metals, Cu and Zn

177	were of most concern because concentrations of both metals exceeded (with Cu 6 out of 7 sediments;
178	with Zn 5 out of 7 sediments) the sediment quality assessment guideline (SQAGs) threshold effect
179	levels (TELs) for coastal and marine waters set by the Florida Department of Environmental Protection
180	(FDEP) for Cu (18.7 mg/kg) and Zn (124 mg/kg, dw) (MacDonald et al.1996). Cu concentrations of the
181	6 sediments also exceeded the informal quick screening marine sediment value (effect range low (ERL)
182	concentration = 34 mg/kg) used for the NOAA SQuiRTs (Buchman, 1999; Long et al. 1995). Zn
183	concentrations were equal to or greater than the ERL (150 mg/kg) for 3 sediments. Neither copper nor
184	zinc concentrations exceeded the NOAA SQuiRTs ERM (effects range median) values which are
185	representative of concentrations above which effects frequently occur (Cu ERM = 270 mg/kg; Zn ERM
186	= 410 mg/kg). Concentrations of the other toxic metals (e.g., As, Cd, Cr, Ni, Pb) in the sediments were
187	not detected or were less than the FDEP TEC SQAGs and the NOAA SQuiRTs ERLs. In a sediment
188	survey in 1982 from the St. Lucie Estuary (SLE) the mean concentration of Cu and Zn were 41
189	(maximum: 229 mg/kg) and 67 (maximum: 235 mg/kg) mg/kg, respectively (Haunert, 1988). Metal
190	concentrations were related to particle size and organic content; as the quantity of clay- and silt-sized
191	particles increased the concentrations of these metals increased. The sediment in the central part of the
192	North Fork of the SLE had the highest concentrations of organic material with the highest
193	concentrations of metals. In 1992, a sediment survey in the Indian River Lagoon, system also showed
194	high concentrations of Cu above background in sediments and clams (Trocine and Trefry, 1993, 1996).
195	In a follow-up sediment survey in 2006-2007, the mean concentrations of Cu (44mg/kg; maximum: 162
196	mg/kg) and Zn (95 mg/kg; maximum: 231 mg/kg) increased from the 1992 survey (Trefry and Trocine,
197	2011).

198 Minimal concentrations of chlorinated organic pollutants were detected in the sediments (7) 199 from the sites except for DDT metabolites (e.g., p.p-DDD) (Table 2b). The total concentrations of DDTs in 3 out of the 7 sediments were higher than the NOAA SQuiRTs ERL concentration (1.58 mg/kg, dw). 200 At the end of the *M. mercenaria* bioconcentration study, sediments samples were collected for 201 Cu and Zn analysis. In general, Zn and Cu concentrations at the beginning (day 0) and the end (day 28) 202 of the study were not significantly different except for NOA 2581 and 2639, Cu and Zn concentrations 203 at the end of the study appeared to be higher than those at the beginning (Table 3). This result indicates 204 that Cu and Zn did not desorb to the overlying water. The high percent of silt, clay, and organic matter 205 206 in the sediments explains why little Cu and Zn release occur during the study. Zn and Cu concentrations in the sediments were also significantly correlated, revealing that Zn and Cu would come from the same 207 208 source.

209

## 210 Water quality and chemistry

Water quality conditions for both studies were within U.S. EPA test guideline requirements. For 211 the A. abdita test, the 10-d average temperature, DO, pH, and salinity of the overlying water during the 212 test were  $21 \pm 1^{\circ}$  C,  $7.1 \pm 1.2$  mg/L,  $8.26 \pm 0.07$ , and  $30 \pm 1$  ppt, respectively. Ammonia concentration 213 214 ranged from 0.4 to 1 mg/L which were less than the U.S. EPA criteria at a pH of 8.26 (3.4 mg/L). Concentrations of dissolved Cu in the overlying and pore water were at the background level (6 µg/L Cu 215 in saltwater used for testing). Concentration of dissolved Zn in the overlying and pore waters ranged 216 217 from the background level (6  $\mu$ g/L Zn) to 33 $\mu$ g/L Zn. These results may be explained by the high percent of silt, clay, and organic matter in the sediments, resulting in negligible desorption of Cu and Zn 218 from the sediments to water. These results also suggest low bioavailability of Cu and Zn in pore water. 219 220 Concentrations of dissolved organic matter in the overlying water were low (< 5mg/L). For the M.

*mercenaria* study, the 28-d average temperature, DO, pH, and salinity of the overlying water were  $24 \pm 0.1^{\circ}$  C,  $6.6 \pm 0.2$  mg/L,  $8.24 \pm 0.11$ , and  $30 \pm 1$  ppt, respectively. Similar to the *A. abdita* study, concentrations of dissolved Cu and Zn in the overlying and pore water were at the background levels, suggesting low Cu and Zn bioavailability. Concentrations of DOM in the overlying water were also low (< 4mg/L).

226

# 227 Effects on survival, Cu and Zn uptake, and growth

Since concentrations of Cu and Zn exceeded both the Florida sediment quality guidelines and the 228 229 NOAA SQuiRTs and the other toxic metals (e.g., As, Cd, Cr, Ni, Pb) were below these numerical values, the discussion in this section considers only Cu and Zn. Results of organism survival are shown 230 in Table 4. Mortality of A. abdita and M. mercenaria ranged from 14% (NOA2333) to 28% (NOA2581) 231 232 and 0% (NOA2639, NOA2570) to 4% (NOA2569), respectively. In general, there was no significant difference between mortality of the tested organisms for the field-collected reference and contaminated 233 sediments. Although the results of SEM and AVS indicate metal bioavailability, the high organic matter 234 235 content in the sediments most likely decreased Cu and Zn bioavailability and toxicity. No mortality was reported in a similar study conducted by Rule (1985) with M. mercenaria and sediments collected from 236 237 the Port of Hampton Roads, Virginia which had similar total sediment concentration of Zn, Pb, Ni, and Cu  $(3 \mu mol/g)$  compared to the present study. 238

239 Concentrations of Cu and Zn in *M. mercenaria* tissue ranged from 9 (background) to 35 mg/kg

dw and 102 (background) to 271 mg/kg dw, respectively (Table 5). In general, Cu and Zn

concentrations in *M. mercenaria* tissue were higher on days 3 through 28 than the background

concentrations (day 0). In addition, sediment Cu and Zn concentrations were positively correlated with

tissue Cu and Zn concentrations (Table 9). These results indicate that *M. mercenaria* accumulated Zn

and Cu from the sediments. As discussed above, results of the sediment chemistry (e.g., total SEM/AVS
> 1) indicate metal bioavailability. This might explain the Cu and Zn accumulation in *M. mercenaria*.
Rule (1985) also found that *M. mercenaria* accumulated Zn from the sediments which had a similar Zn
sediment concentration to the present study. However, Zn accumulation by *M. mercenaria* in the Rule
(1985) study was approximately half the Zn accumulation in the present study but Zn bioavailability in
the present study was higher than in the Rule (1985) study.

Cu and Zn concentrations in *M. mercenaria* tissue in the present study did not increase over time. 250 This is in contrast with our earlier studies with freshwater Florida apple snails (*Pomacea paludosa*) 251 252 where apple snails accumulated Cu from sediment overtime (Hoang et al. 2008b, Hoang et al. 2011). Cu and Zn concentrations in *M. mercenaria* shell were also below the detection limits but in general, 253 negligible Cu and Zn concentrations were detected in apple snail shells as well (Hoang et al. 2008b). 254 Zn and Cu concentrations in A. abdita varied from site to site and ranged from 111 (NOA2569) to 255 355 mg/kg dw (NOA2581) and 65 (background) to 364 mg/kg dw (NOA2570), respectively (Table 6). 256 In general, Cu concentrations in A. abdita were higher at the end of the study than at the beginning of 257 the study for all sites. There was a positive correlation between the A. abdita tissue Cu concentration 258 259 and sediment Cu concentration (Table 10). Results of the present study indicate that A. abdita 260 accumulated Cu from the sediments during the 10 day study. Results of the sediment chemistry suggest Cu bioavailability and therefore explain the Cu accumulation in A. abdita. The final tissue results 261 obtained for A. abdita in the present study were similar to the tissue Cu concentrations found in our 262 263 earlier 10 day study with the freshwater benthic amphipod, Hyalella azteca exposed to Cu-contaminated soils from citrus agricultural sites near the St. Lucie watershed (Hoang et al. 2009b). Hoang et al. 2009b 264 showed that Cu tissue (whole body) concentrations ranged from 128 to 294mg/kg after 10 days 265 266 exposure when initial Cu soil concentrations ranged from 5-234 mg/kg from these citrus agricultural

sites. Note that in this *A. abdita* study, Cu tissue concentrations were up to 6 times the initial Cu tissue
concentrations following only 10 days of exposure and similar to the response of the freshwater
amphipod, *H. azteca* following 10 day exposures. Furthermore, the tissue results for both freshwater and
saltwater benthic species, following exposures to Cu in sediment, raise some interesting issues for
burrowing and tube dwelling in faunal benthic species which have habitats in close contact with
sediment (pore water) for part or most of their life cycle. The influence on Cu uptake in these benthic
species on upper trophic level diets and species has not been extensively investigated.

Also note that accumulated metal within and between invertebrate taxa, vary considerably, even 274 275 in the absence of anthropogenic pollution (Rainbow, 2002). For trace metals like zinc and copper, which play essential roles in metabolism of most invertebrate, the quantity necessary to perform these 276 functions may also vary widely within and between invertebrate taxa. For aquatic organisms used in 277 toxicity tests, it is also critical to know the background holding, culture and water quality conditions 278 because if test organisms are obtained from different aquaculture sources the trace metal concentrations 279 in their tissues and organs may be different and will obviously reflect prior water quality and diet they 280 281 were exposed to. Background tissue concentrations (day 0) of zinc (Table 6) for A. abdita were as high as concentrations at the end of the 10d treatment and therefore precluded any comparisons of zinc tissue 282 283 concentrations. In addition, to being cautious when obtaining organisms from aquaculture sources for aquatic toxicity testing, the use of field-collected species requires even greater awareness for use in 284 toxicity testing especially for the evaluation of hazards and risks. 285

The dry weight of *M. mercenaria* and *A. abdita* are shown in Tables 7 and 8, respectively. Dry weight of *M. mercenaria* shell, tissue, and whole body from days 0-28 ranged from 194 to 355 mg/organism, 5 to 8 mg/organism, and 198 to 552 mg/organism, respectively. In general, dry weight was not affected by sediment exposure up to day 21. However, there was a slight negative correlation 290 between *M. mercenaria* tissue weight and Cu and Zn concentrations in the sediments at the end of the 291 study (day 28) (Table 9). This suggests that Cu and Zn concentrations in the sediments may have started to produce an effect on *M. mercenaria* growth (as measured by dry weight) after 28 days exposure and 292 that in the experimental design the uptake (exposure) phase was too short and it should have been 293 extended. Clams that live in these sediments will be continuously exposed to Cu and Zn for most of 294 295 their life cycle in the in the St. Lucie system, with little time for depuration and recovery, hence the effect might be greater. Results of the present study thus raise concern about the population dynamics of 296 bivalves in the St. Lucie ecosystem. 297

298 The St. Lucie estuarine ecosystem has been documented as a Cu-contaminated system for decades (Long and Morgan 1990, Delfino et al. 1991, FDEP 1994). The cause of Cu contamination is likely due 299 to Cu release from the sandy soils of nearby citrus agriculture farms through surface runoff. Cu will 300 continuously be used in citrus agriculture, with a long season of application thus increasing the Cu load 301 and release from soils into runoff water, adding more Cu into the St. Lucie estuary and the Everglades 302 ecosystems (Hoang et al. 2008a, Hoang et al. 2009a). This is also based on an exposure analyses of 303 copper concentrations in water and sediment in south Florida aquatic systems for over 15 years from 304 1990-2008, which shows high Cu concentrations in aquatic systems and potential risks to mollusks in 305 306 the North and South forks of the SLR and in the SLE (Carriger and Rand, in press).

307 Dry weight of *A. abdita* was negatively correlated with tissue Cu concentration (Table 10). This 308 suggests that tissue Cu concentrations affected *A. abdita*'s growth. Since water concentrations of Cu 309 and Zn were at typical background levels, the effects of Cu and Zn on *M. mercenaria* tissue weight and 310 *A. abdita* weight would be due to exposure via sediment ingestion. Furthermore, tissue Cu and Zn 311 concentrations were correlated with both sediment Cu and Zn concentrations, suggesting that Cu and Zn 312 simultaneously entered the organisms. Ingestion of Cu- and Zn-bound to organic matter in sediments was thus a major exposure route. Several studies have demonstrated that metal exposure to clams and
amphipods via food and sediment ingestion was more important than pore water exposure (Eriksson and
Sundelin 2002, Labreche et al. 2002, Forbes et al. 1998).

316

#### 317 Summary and conclusions

The present study showed that sediments collected from the St. Lucie estuarine system contained 318 Cu and Zn concentrations that exceeded both the Florida State sediment quality criteria and NOAA 319 SQuiRTs sediment values. The total concentration of DDTs in 3 out of the 7 sediments was also higher 320 321 than the NOAA SQuiRTs sediment values. M. mercenaria and A. abdita exposed to the St. Lucie sediments resulted in Cu accumulation in A. abdita and Cu and Zn accumulation in M. mercenaria. The 322 present study also indicated that Cu and Zn exposures via sediment ingestion were most likely an 323 important route of exposure. However, there was no effect of the contaminated sediments on organism 324 survival. 325

Elevated Cu and Zn concentrations in the tissues and the effects on the weight of both M. 326 *mercenaria* and *A. abdita* raise concerns for the long-term viability of invertebrate populations, for 327 higher trophic organisms in the St. Lucie estuarine ecosystem and the population dynamics of the 328 329 ecosystem because these are only two organisms that are primary food resources in the St. Lucie system which are exposed to the contaminated sediments for either part or their entire life cycles. The 330 significance of these results can only be fully realized when studies are conducted with other organisms 331 332 exposed to a greater number field-collected sediments from a larger number of sediment sites. These studies gain in importance in lieu of the continued input of these metals into the environment. 333

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- 335

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