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α-Alkyl-α-Amino-β-Sulphone Hydroxamates as Potent MMP Inhibitors Which Spare MMP-1

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Abstract: A series of α-alkyl-α-amino-β-sulphone hydroxamates was prepared and evaluated for potency versus MMP-2 and MMP-13, and for selectivity versus MMP-1. Low nanomolar potency was obtained with selectivity versus MMP-1 ranging from >10 to >1,000. Selected compounds were orally bioavailable.

Introduction

In our previous letter we described a series of α-amino-β-sulphone hydroxamates which are potent inhibitors of MMP-13 which spare MMP-1. Overexpressed MMPs play a crucial role in tumor growth and metastasis in cancer, and in the destruction of articular cartilage in osteoarthritis (OA) and rheumatoid arthritis (RA). Hence, the inhibition of the relevant MMP enzymes may prove to be clinically effective in halting the advance of these diseases. The gelatinases A and B (MMP-2 and MMP-9) have been implicated in tumor progression, and MMP-13 has been implicated in the destruction of articular cartilage in arthritis. Herein we report the preparation and preliminary SAR of a series of α-amino-α-alkyl-β-sulphones which are highly selective in sparing MMP-1, based on the hypothesis that the musculoskeletal side effect observed clinically with the broad-spectrum inhibitor marimastat is due to potent inhibition of MMP-1. Alkyl substituents alpha to the hydroxamate were employed to modulate pharmacokinetic properties including absorption and half-life, as well as physicochemical properties, while the P1 substituent was varied to optimize potency and selectivity.

Chemistry

The targeted α-methyl-α-amino diphenyl ether hydroxamates of type 5 were prepared starting with a halide displacement of chloroacetone (1) with 4-phenoxythiophenol to afford 2 (Scheme 1). Strecker synthesis then gave the nitrile 3 which was hydrolyzed to the carboxylic acid. Oxidation of the sulfide after protection of the amino group as the acetamide, and subsequent deprotection then gave the amino acid 4. Functionalization of the amino group was accomplished by alkylation, acylation or reductive amination as required, and standard EDC coupling with hydroxylamine afforded the diphenyl ether sulphone hydroxamates 5.
The diaryl thioether 7 was prepared by reacting chloroacetone (1) with 4-fluorothiophenol (R = F). Strecker synthesis gave nitrile 3 (R = F) which was hydrolyzed and protected as the acetamide (Scheme 2). Oxidation then afforded the corresponding 4-fluorophenylsulphone, and nucleophilic aromatic substitution with thiophenol gave carboxylic acid 6. Acetamide hydrolysis preceded amino functionalization with the appropriate R² reagent, and EDC coupling afforded the hydroxamates 7.

As illustrated in Scheme 3, α-pyrrolidine-β-sulphones were prepared from racemic methyl N-Cbz-proline (8). Alkylation with methylene diiodide gave the α-iodomethyl derivative which was used to alkylate 4-phenoxythiophenol, and oxidation gave the sulphone 9. The Cbz protecting group was removed by hydrogenolysis, exposing the amine which was functionalized by alkylation with propargyl bromide. Saponification of the methyl ester and coupling with hydroxylamine then afforded the hydroxamate 10 (R² = propargyl).

The α-phenyl-α-amino derivative 15 (Scheme 4) was prepared from D,L-phenylglycine (11) by benzylation and treatment with acetic anhydride to give the 2-phenyloxazolone 12. Alkylation of this azlactone with methylene diiodide gave iodomethyl azlactone 13, and displacement of the iodide with 4-phenoxythiophenol and subsequent oxidation with metachloroperbenzoic acid gave the sulphone 14.
oxazolone was then hydrolyzed and the resulting carboxylic acid coupled with TMS-protected hydroxylamine to afford 15 (R² = H). Alternatively, the oxazolone ring was opened directly with hydroxylamine to afford the benzamide hydroxamate (R² = Bz).

**Scheme 4**

![Scheme 4](image)

**Results and Discussion**

Table 1 summarizes the potency versus MMP-2, MMP-13 and MMP-1 for compounds of generic structures 5, 7 and 15. The diaryl ethers (X = O) were an order of magnitude more potent than the corresponding thioethers (X = S), although the thioethers were noted to be somewhat more selective in sparing MMP-1 (5a and 5b versus 7a and 7b). Amides of the α-amino group (compounds 5a, 5j, 7a and 15a) were not well tolerated, whereas simple alkyl and aralkyl amines were potent for both MMP-2 and MMP-13. Disubstitution on the amine led to a loss of potency (5c). The α-phenyl amine 15b was potent for MMP-13 and MMP-2, but was also somewhat more potent for MMP-1. Almost all compounds exhibited excellent selectivity versus MMP-1, in several cases exceeding 1000X for the ratio of IC₅₀ values (MMP-1/MMP-2 and MMP-1/MMP-13), in contrast to the broad-spectrum inhibitors CGS 27023A and marimastat.

Table 2 shows the enzyme potency of proline-derived analog 10. Since the racemate 10 was found to be quite potent, the material was resolved into its enantiomers via chiral chromatography. The first eluter, hydroxamate 10a, was found to be the more potent enantiomer (eutomer) by at least two orders of magnitude against both MMP-13 and MMP-2 as compared to the less potent enantiomer (distomer) 10b. Compound 10a was also highly selective in sparing MMP-1 (3000X).
**Table 1:** IC$_{50}$ (nM)$^{10}$ values for α-alkyl-α-amino-β-sulphone hydroxamates

<table>
<thead>
<tr>
<th>Compound</th>
<th>$R^1$</th>
<th>$R^2$</th>
<th>$R^3$</th>
<th>X</th>
<th>MMP-13</th>
<th>MMP-2</th>
<th>MMP-1</th>
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<tbody>
<tr>
<td>5a</td>
<td>CH$_3$</td>
<td>Ac</td>
<td>H</td>
<td>O</td>
<td>41.5</td>
<td>40.0</td>
<td>&gt;10,000</td>
</tr>
<tr>
<td>5b</td>
<td>CH$_3$</td>
<td>H</td>
<td>H</td>
<td>O</td>
<td>0.2</td>
<td>0.6</td>
<td>170</td>
</tr>
<tr>
<td>5c</td>
<td>CH$_3$</td>
<td>CH$_3$</td>
<td>CH$_3$</td>
<td>O</td>
<td>24.0</td>
<td>5.0</td>
<td>&gt;10,000</td>
</tr>
<tr>
<td>5d</td>
<td>CH$_3$</td>
<td>CH$_2$CH$_3$</td>
<td>H</td>
<td>O</td>
<td>1.6</td>
<td>1.3</td>
<td>1600</td>
</tr>
<tr>
<td>5e</td>
<td>CH$_3$</td>
<td>CH$_2$Ph</td>
<td>H</td>
<td>O</td>
<td>0.3</td>
<td>0.2</td>
<td>1200</td>
</tr>
<tr>
<td>5f</td>
<td>CH$_3$</td>
<td>CH$_2$CH$_2$Ph</td>
<td>H</td>
<td>O</td>
<td>2.4</td>
<td>1.3</td>
<td>2400</td>
</tr>
<tr>
<td>5g</td>
<td>CH$_3$</td>
<td>3,4-methylenedioxybenzyl</td>
<td>H</td>
<td>O</td>
<td>1.1</td>
<td>0.5</td>
<td>2350</td>
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<tr>
<td>5h</td>
<td>CH$_3$</td>
<td>2-naphthylmethyl</td>
<td>H</td>
<td>O</td>
<td>1.4</td>
<td>0.4</td>
<td>&gt;10,000</td>
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<tr>
<td>5i</td>
<td>CH$_3$</td>
<td>propargyl</td>
<td>H</td>
<td>O</td>
<td>0.6</td>
<td>0.2</td>
<td>700</td>
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<tr>
<td>5j</td>
<td>CH$_3$</td>
<td>pyrrolidineacetyl</td>
<td>H</td>
<td>O</td>
<td>160</td>
<td>80</td>
<td>&gt;10,000</td>
</tr>
<tr>
<td>7a</td>
<td>CH$_3$</td>
<td>Ac</td>
<td>H</td>
<td>S</td>
<td>580</td>
<td>540</td>
<td>&gt;10,000</td>
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<tr>
<td>7b</td>
<td>CH$_3$</td>
<td>H</td>
<td>H</td>
<td>S</td>
<td>2.4</td>
<td>3.2</td>
<td>4400</td>
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<tr>
<td>15a</td>
<td>Ph</td>
<td>benzoyl</td>
<td>H</td>
<td>O</td>
<td>161</td>
<td>184</td>
<td>&gt;10,000</td>
</tr>
<tr>
<td>15b</td>
<td>Ph</td>
<td>H</td>
<td>H</td>
<td>O</td>
<td>0.4</td>
<td>0.2</td>
<td>130</td>
</tr>
</tbody>
</table>

**CGS 27023A**

5.1  4.6  34.3

**marimastat**

2.0  0.75  2.9

**Table 2:** IC$_{50}$ (nM)$^{10}$ values for α-pyrrolidine-β-sulphone hydroxamates

<table>
<thead>
<tr>
<th>Compound</th>
<th>$R^2$</th>
<th>MMP-13</th>
<th>MMP-2</th>
<th>MMP-1</th>
</tr>
</thead>
<tbody>
<tr>
<td>10  (racemic)</td>
<td>propargyl</td>
<td>1.3</td>
<td>0.2</td>
<td>400</td>
</tr>
<tr>
<td>10a (eutomer)</td>
<td>propargyl</td>
<td>&lt;0.1</td>
<td>&lt;0.1</td>
<td>300</td>
</tr>
<tr>
<td>10b (distomer)</td>
<td>propargyl</td>
<td>60.0</td>
<td>19.3</td>
<td>&gt;10,000</td>
</tr>
</tbody>
</table>
Selected analogs were dosed orally in rats at 20 mpk to assess absorption by measuring $C_{\text{max}}$, and the concentration remaining at 6 hours was used as an initial rough indicator of the half-life. The $\alpha$-methyl-$\alpha$-amino analog $5b$ showed a high $C_{\text{max}}$ of 6.43 ug/ml, somewhat greater than the corresponding thioether $7b$ ($C_{\text{max}} = 1.54$ ug/ml). N-ethyl and N-benzyl analogs $5d$ and $5e$ were moderately well absorbed ($C_{\text{max}} = 0.561$ and 0.216 ug/ml, respectively), and N-propargyl amine $5i$ exhibited a $C_{\text{max}}$ of 1.37 ug/ml. However, all of the compounds tested were less than 15 ug/ml in plasma at the 6 h time point.

In summary, we have described a promising series of $\alpha$-alkyl-$\alpha$-amino-$\beta$-sulphone hydroxamates which are potent inhibitors of both MMP-2 and MMP-13, and which spare MMP-1. Several analogs showed good absorption when administered orally in the rat. The efficacy of these compounds in animal models of cancer and arthritis will be disclosed in due course.

References and Notes

9. The resolution of 30 was accomplished using a Chiralpak AD column (4.6 mm X 25 cm) at a flow rate of 1.0 ml/min eluting with a mobile phase of 35/65 ethanol/heptane with 0.2% trifluoroacetic acid. Tony Yan is gratefully acknowledged for performing the chiral separation.
10. Inhibitors were assayed against purified hMMP-13, hMMP-1 and hMMP-2 using an enzyme assay based on cleavage of the fluorogenic peptide MCA-Pro-Leu-Gly-Leu-Dpa-Ala-Arg-NH$_2$. This is similar to conditions described by C. G. Knight et. al. in FEBLS Lett. 1992, 296, 263, except that 0.02% final concentration of 2-mercaptoethanol was used in the MMP-13 and MMP-1 assays. All basic compounds were tested as their hydrochloride salts except for 10a and 10b, which were tested as the trifluoroacetate salts.